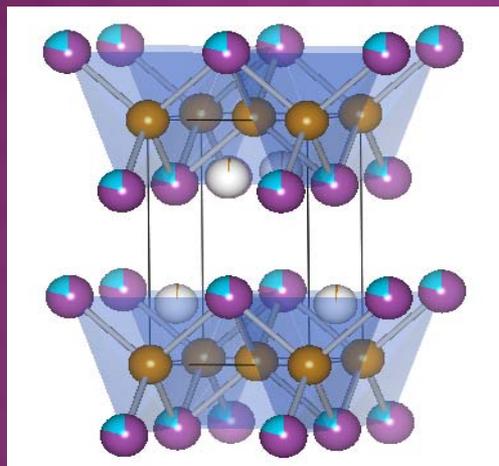


# Neutron Diffraction Studies on the Structure and Chemical Composition of Superconducting Phases of $\text{Fe}_{1+y}\text{Te}_{1-x}\text{Se}_x$



UNIVERSITY OF  
MARYLAND

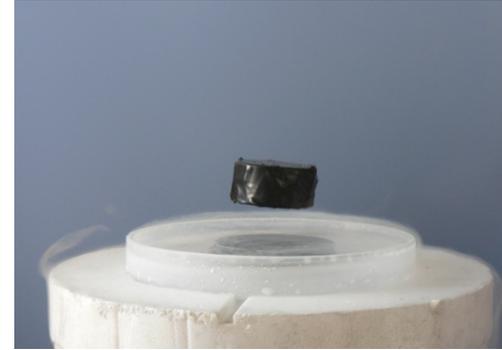
Vikas Bhatia\*, Efrain Rodriguez  
NIST Center For Neutron Research

# Overview

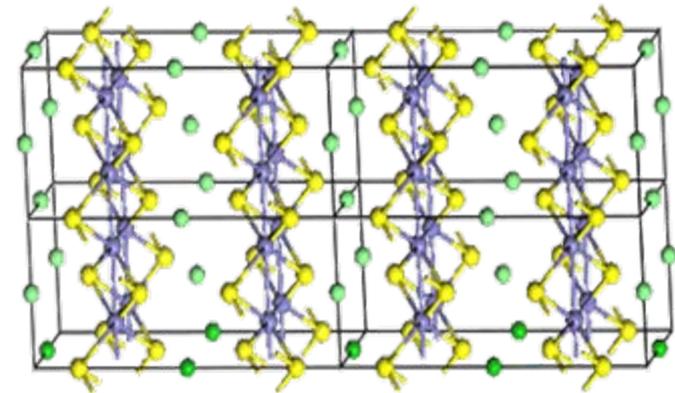
- ◉ Introduction & Motivation
- ◉ X-Ray/Neutron Diffraction
- ◉ Synthesis Methods
- ◉ Rietveld (Structural) Analysis
- ◉ Future Work

# Introduction

- ◉ Superconductivity-  
Electrical Resistivity of  
Zero
- ◉ Iron based compounds  
become superconducting  
at certain chemical  
compositions
  - Iron Pnticides



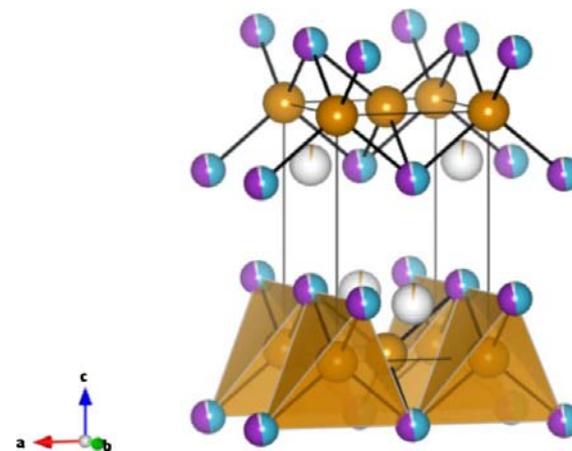
*Meissner Effect*  
Taken from <http://faculty.nwacc.edu>



*Iron Pnticide -CaFe<sub>2</sub>As<sub>2</sub>*  
Taner Yildirim

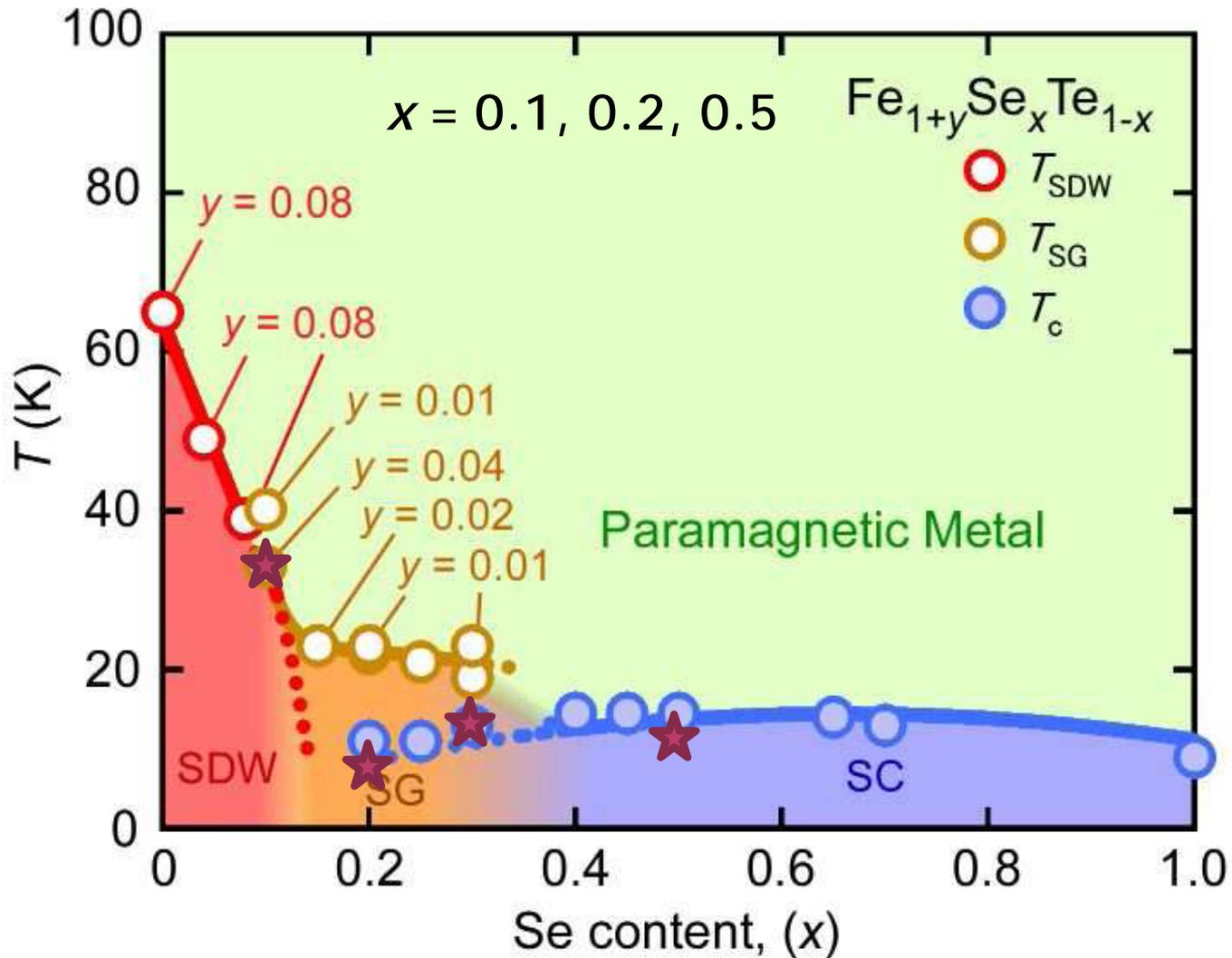
# Motivation

- ◉ Iron Chalcogens have been found to superconduct
- ◉ FeTe becomes superconducting with the doping of Se at certain stoichiometries
  - $\text{Fe}_{1+y}\text{Te}$  - monoclinic or orthorhombic
  - $\text{Fe}_{1+y}\text{Te}_{1-x}\text{Se}_x$  - tetragonal
- ◉ Fe(Te,Se) system is nonconventional
  - Offer the opportunity to study chemical composition, magnetism, and superconductivity



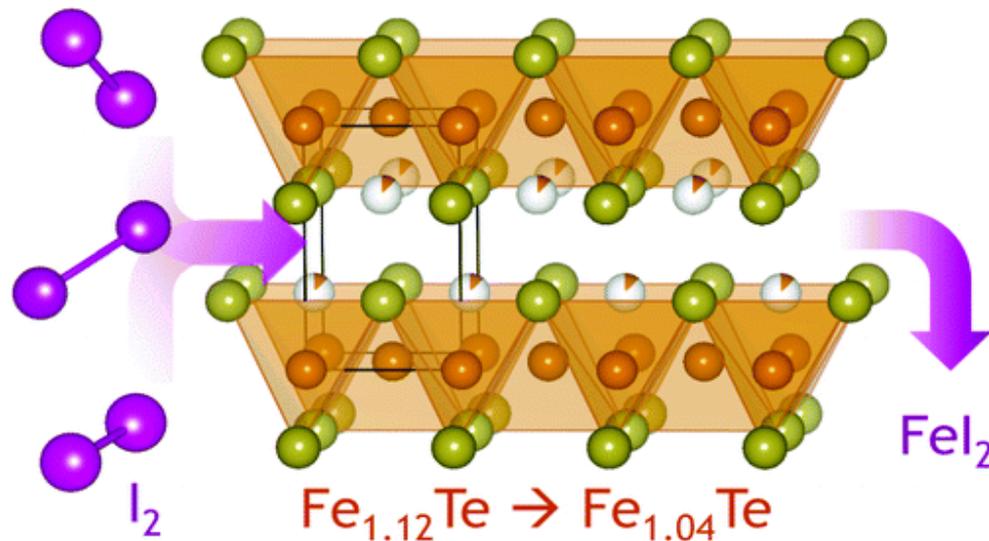
D. Louca, *et al.*, *Phys Rev B.* **2010**, 81

# $\text{Fe}_{1+y}(\text{Te}_{1-x}\text{Se}_x)$ phase diagram- What are we studying?



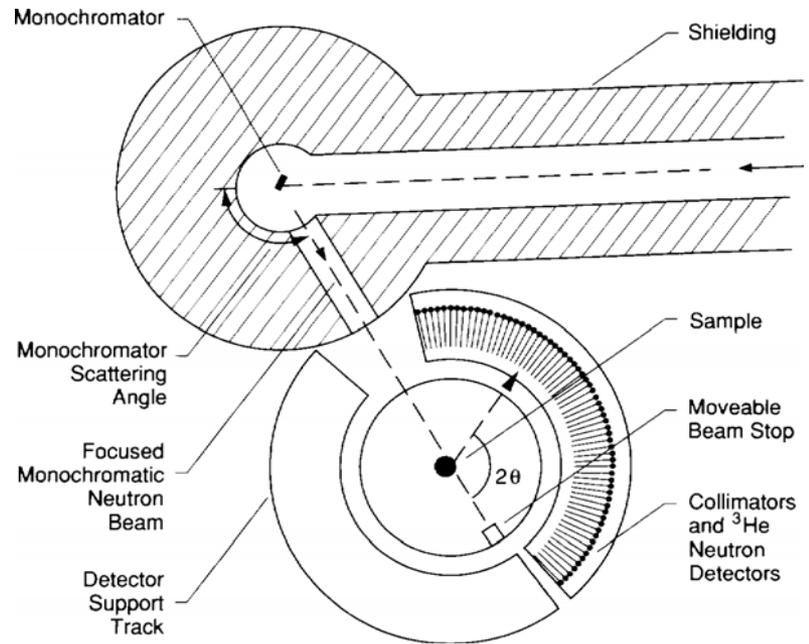
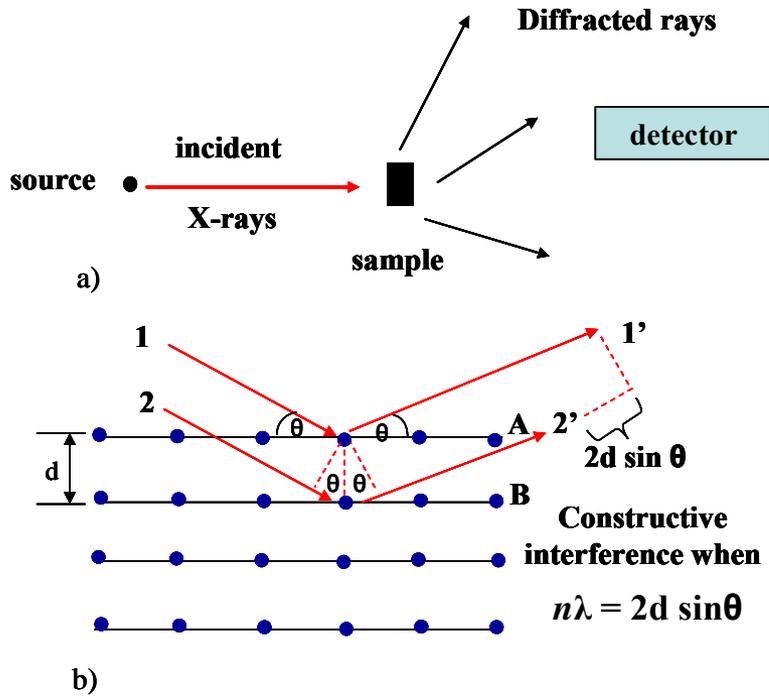
# Soft Chemistry Route- Iodine used to remove excess Iron

Iodine deintercalation to help study the effects of excess iron



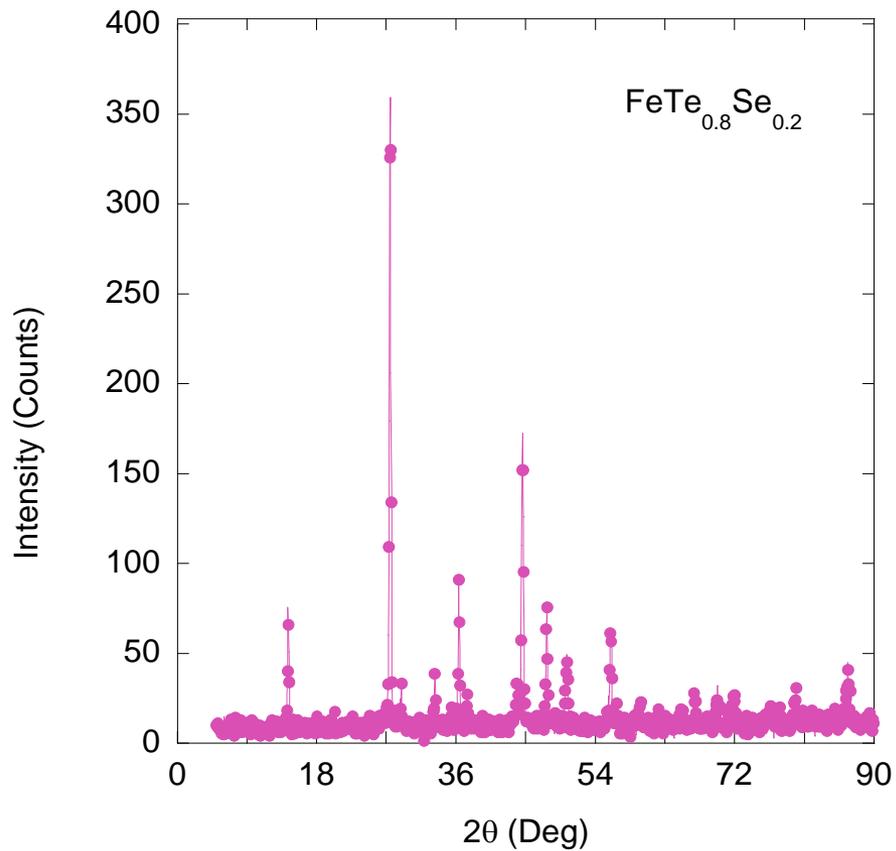
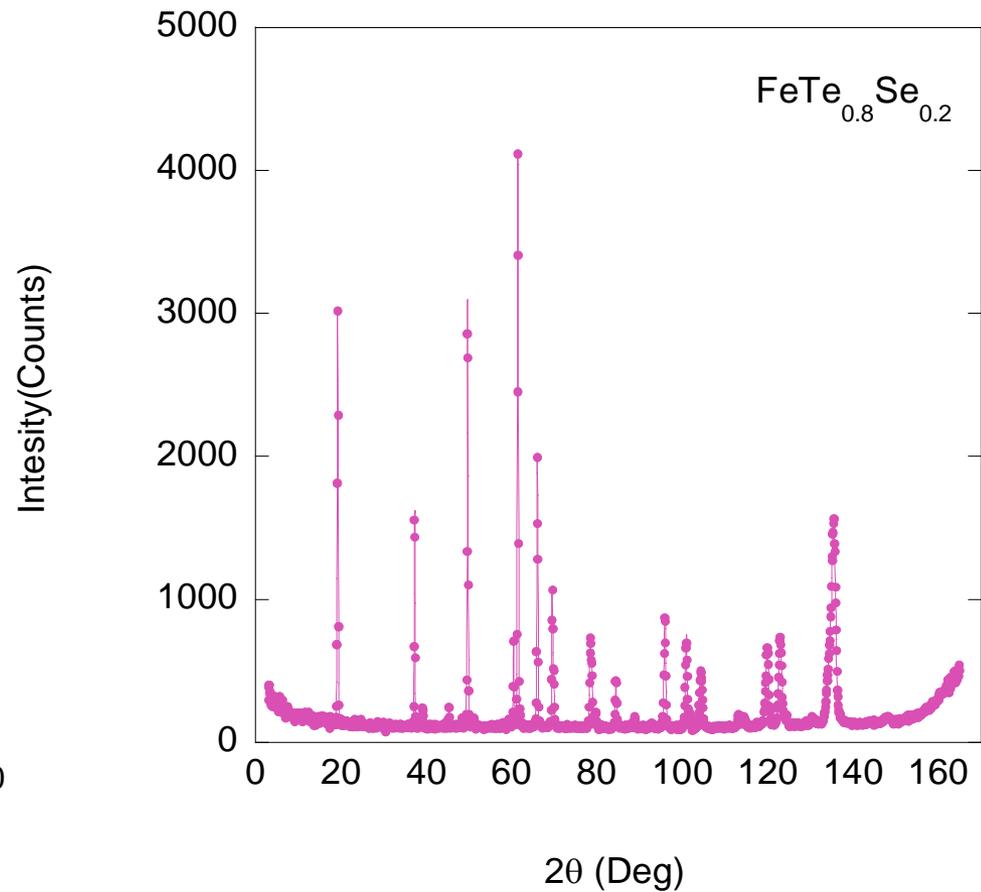
E. E. Rodriguez, *et al.*, *J. Am Chem. Soc.* **2010**, 132

# X-Ray vs. Neutrons



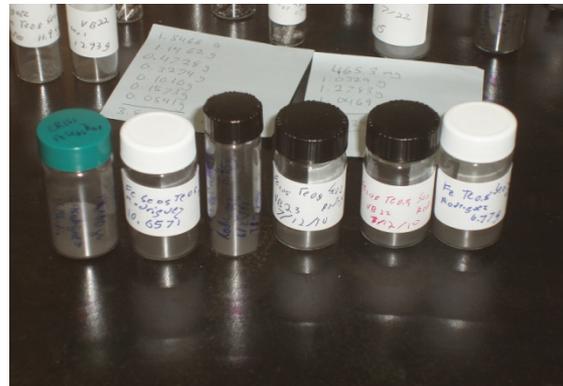
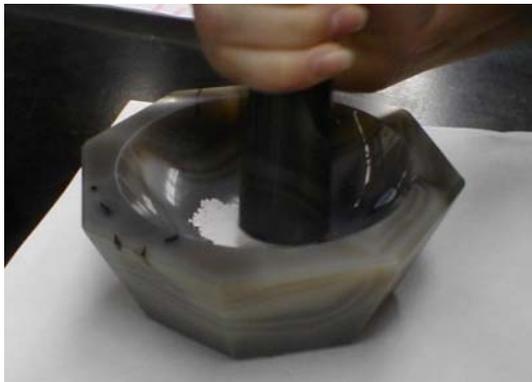
Roger Pynn, "Neutron Scattering: A Primer"

Adapted from: West, A. R., *Basic Solid State Chemistry*. 2<sup>nd</sup> ed.; John Wiley & Sons: New York, 1999, pp. 134-135.

**XRD****Neutron**

# Synthesis — solid state method

- Calculate the amounts of reactants
  - $(1+y)\text{Fe} + (x)\text{Se} + (1-x)\text{Te} \rightarrow \text{Fe}_{1+y}\text{Te}_{1-x}\text{Se}_x$
- Solid State Reaction
- Samples are reacted with pure Iodine vapor to remove excess Iron
  - $\text{Fe}_{1+y}\text{Te}_{1-x}\text{Se}_x + (y)\text{I}_2 \rightarrow \text{FeTe}_{1-x}\text{Se}_x + \text{FeI}_2$
- Sample washed with methanol to remove  $\text{FeI}_2$



# Synthesis — new methods



[1]



Horizontal Bridgman Technique



“Vertical Bridgman Technique”

- New procedure for Iodine reactions was devised
  - Deintercalation of Iron with Iodine/Acetonitrile
- Single crystal growth
  - Horizontal and “Vertical” Bridgeman Technique

# After Samples Are Prepared...



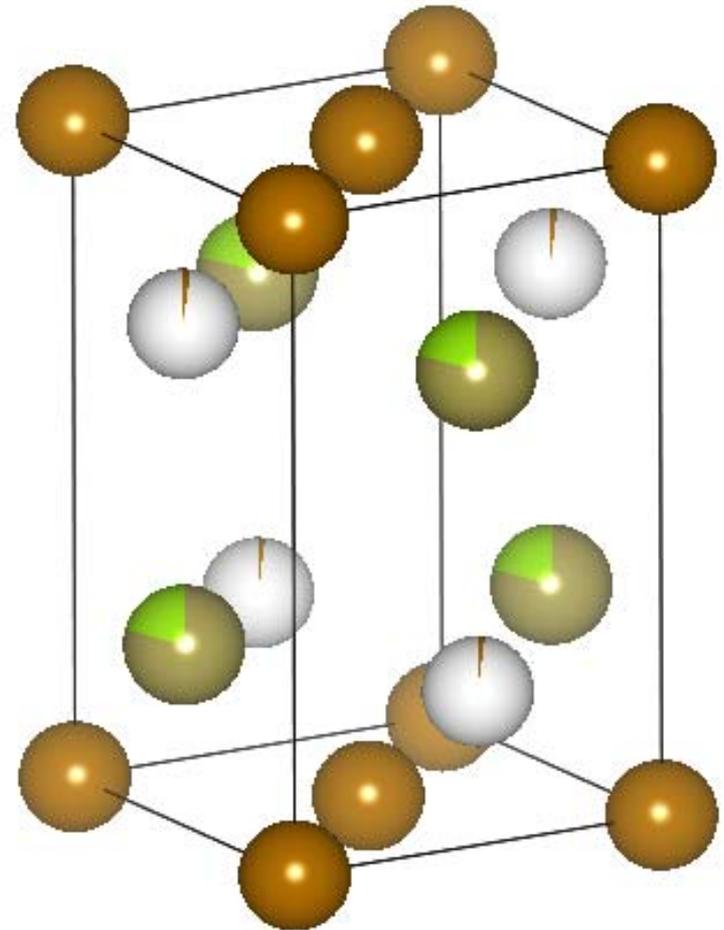
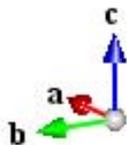
- X-Ray Diffraction
- BT-1
- Rietveld Refinement
- Sent for magnetic measurements



# Rietveld Refinement

- Lattice Parameters
- Occupancies
- Positions
- Bond Distances/ Angles

Iron  
Selenium  
Tellurium



# FeTe<sub>0.8</sub>Se<sub>0.2</sub> Neutron Data RT Before and After I<sub>2</sub> Reaction

Before

After

Interstitial Iron = 3.9(3) %

Tellurium = 80(2) %

Selenium = 20(2) %

a = 3.81024(4) Å

b = 3.81024(4) Å

c = 6.1821(1) Å

Interstitial Iron = 2.3(3) %

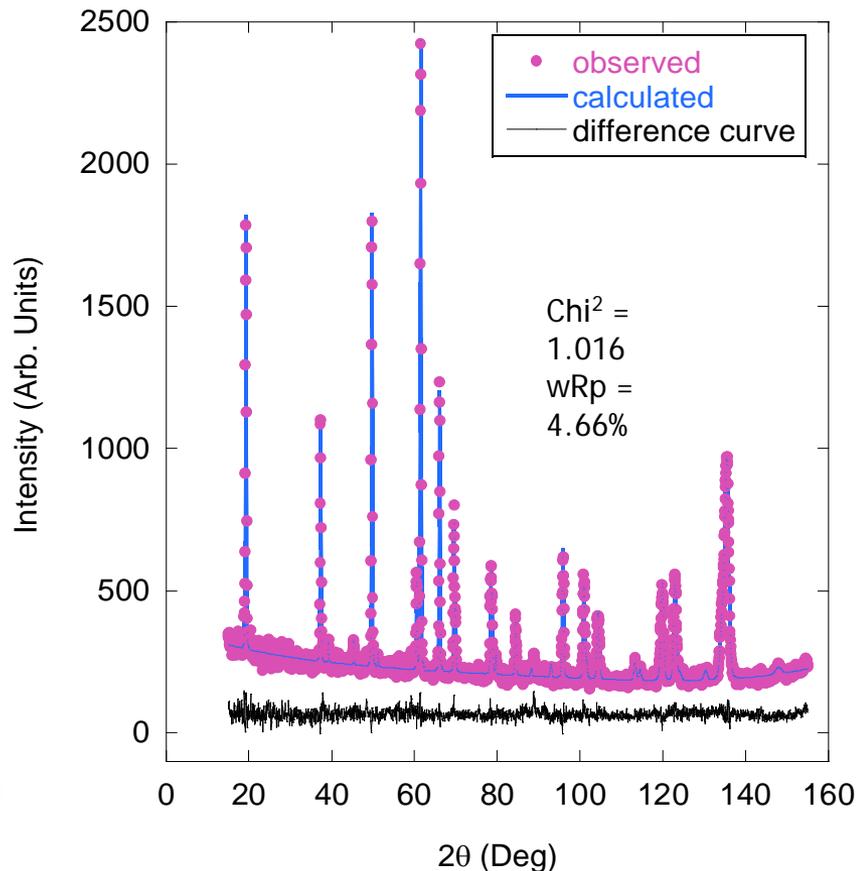
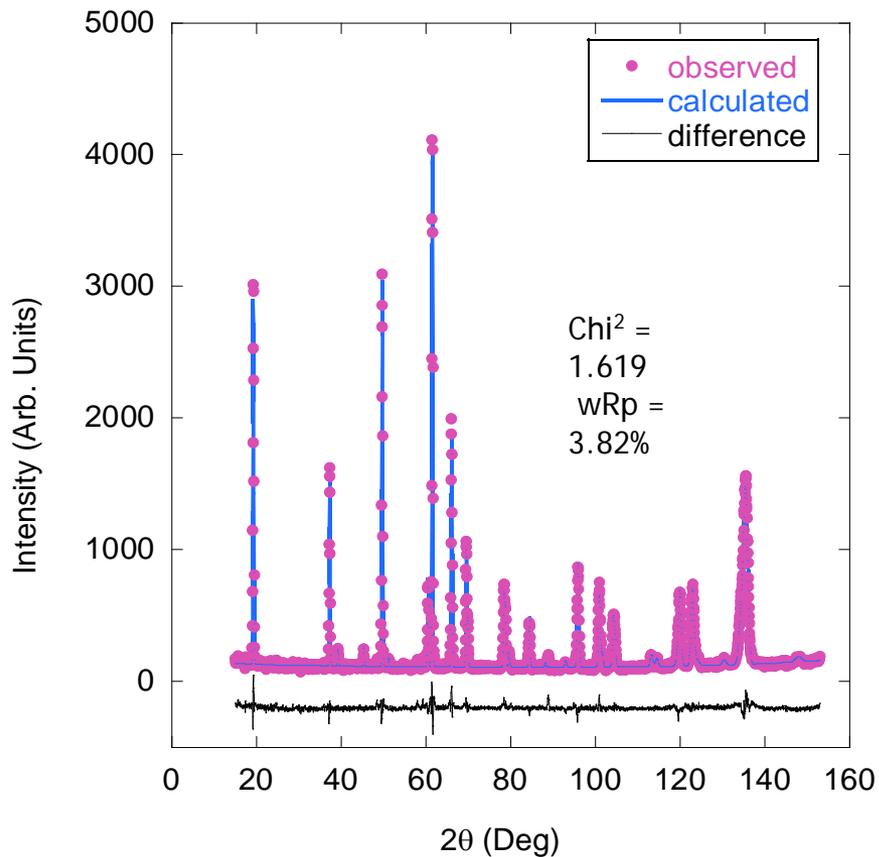
Tellurium = 80(2) %

Selenium = 20(2) %

a = 3.81013(5) Å

b = 3.81013(5) Å

c = 6.18435(1) Å



# FeTe<sub>0.9</sub>Se<sub>0.1</sub> Neutron Data Before I<sub>2</sub> Reactions

- With Selenium decrease
  - Amount of Interstitial Iron increased
  - Lattice distances increased

Interstitial Iron = 5.9(3) %

Tellurium = 92(2) %

Selenium = 8(2) %

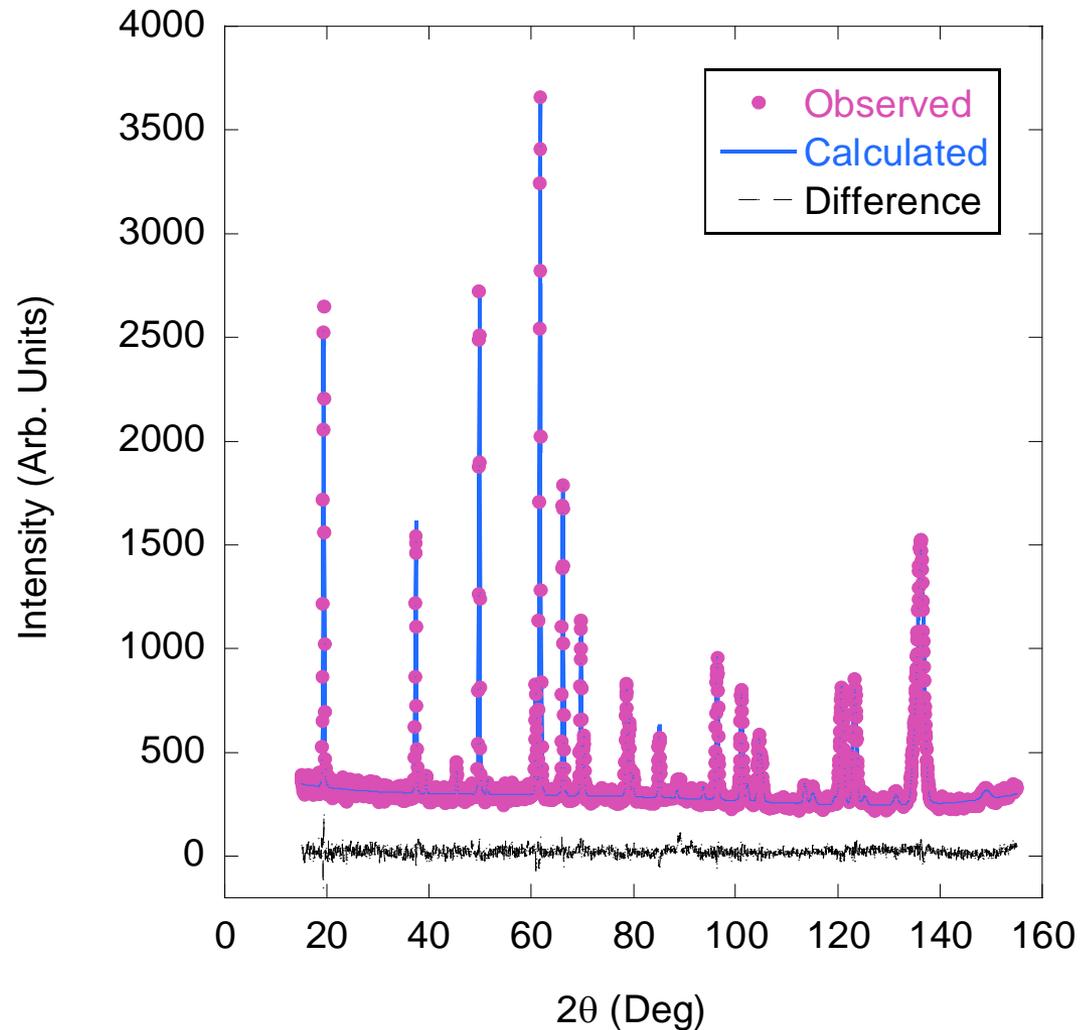
wRp = 7.2%

Chi<sup>2</sup> = 1.284

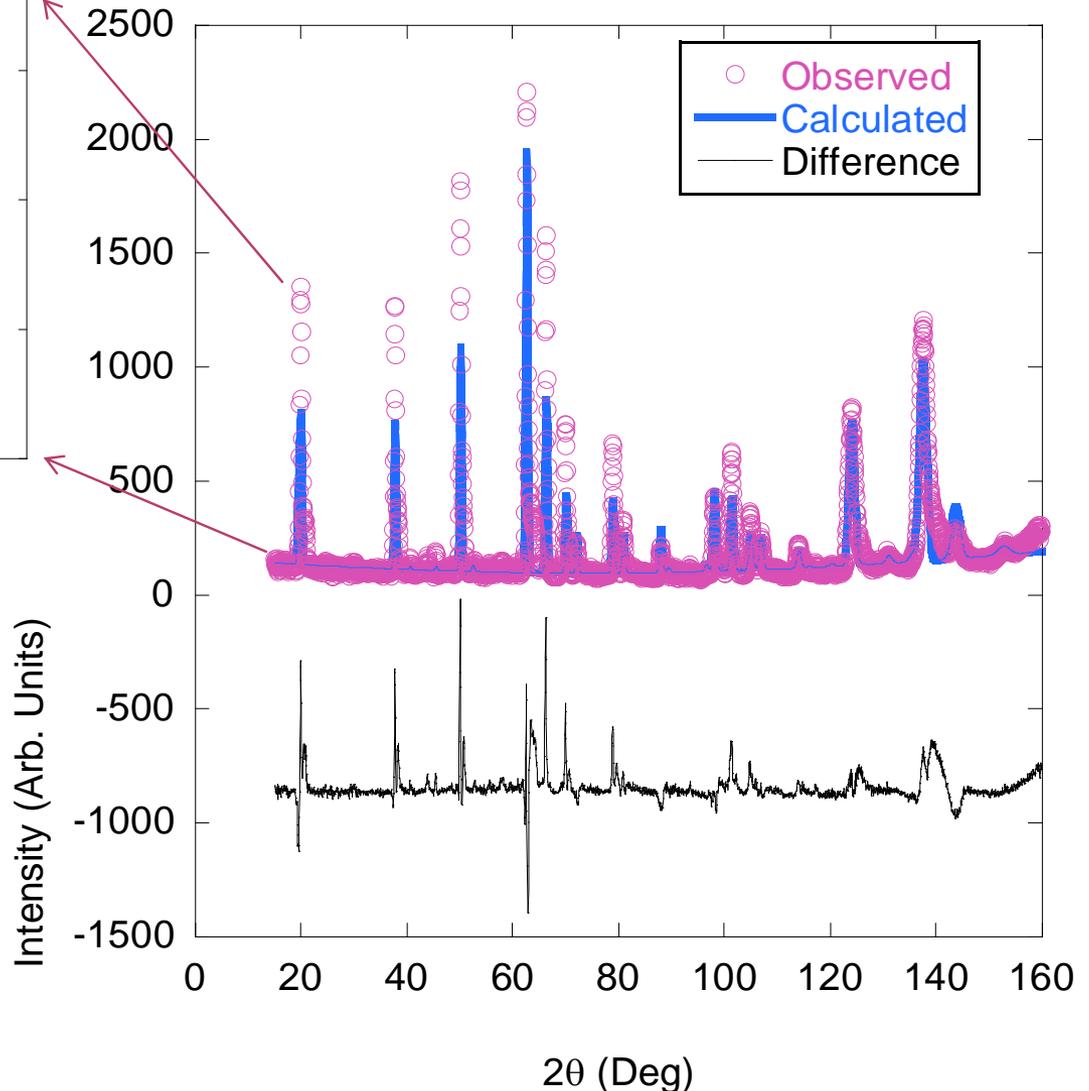
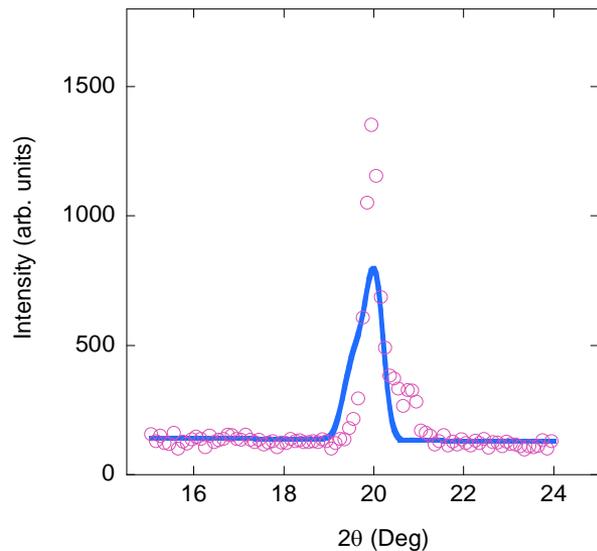
a = 3.8156(1) Å

b = 3.8156(1) Å

c = 6.2367(2) Å



# FeTe<sub>0.5</sub>Se<sub>0.5</sub> Neutron Data 100k Quenched 440° C



Interstitial Iron = 5(1) %

Tellurium = 34(7) %

Selenium = 66(7) %

Chi<sup>2</sup> = 11.5

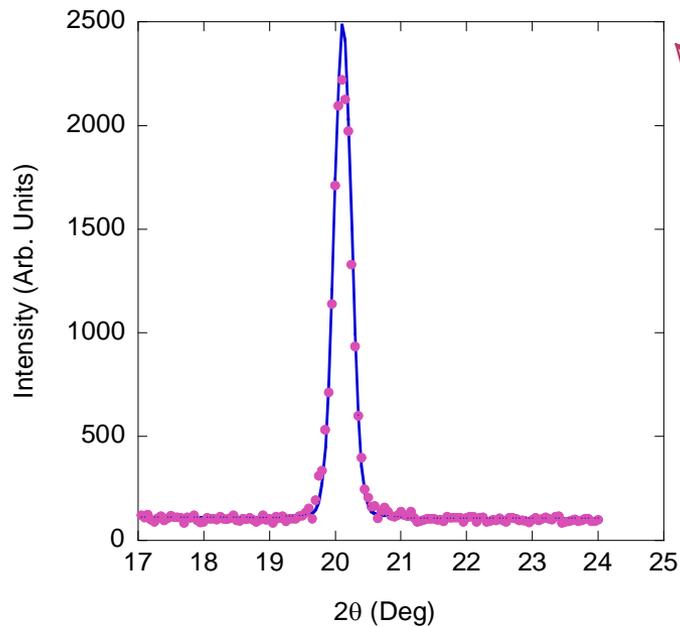
wRp = 28.64%

a = 3.8020(1) Å

b = 3.8020(1) Å

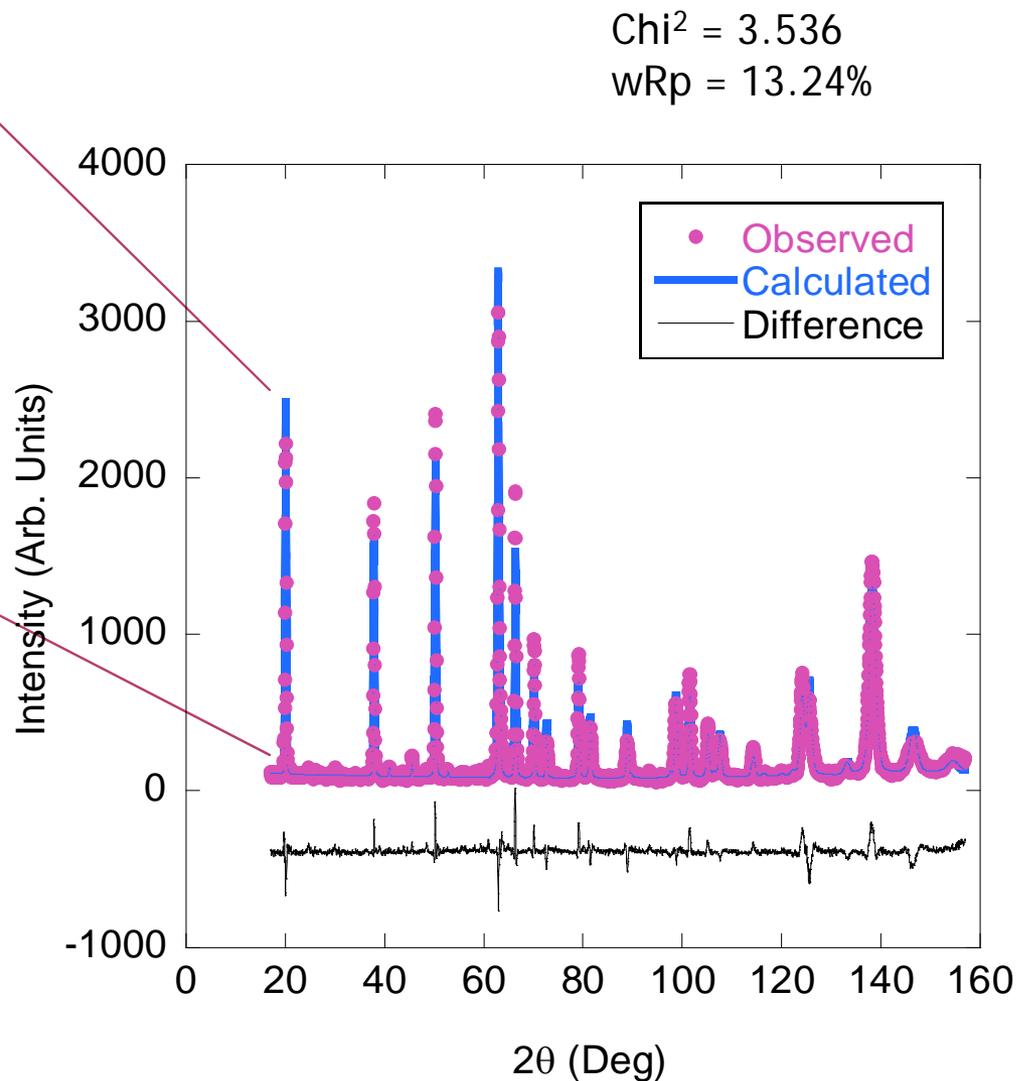
c = 5.9902(5) Å

# FeTe<sub>0.5</sub>Se<sub>0.5</sub> Neutron Data 100k Slow Cooled



Interstitial Iron = 2(1) %  
Tellurium = 52(8) %  
Selenium = 48(8) %  
Chi<sup>2</sup> = 3.51  
wRp = 13.11%

a = 3.7941(1) Å  
b = 3.7941(1) Å  
c = 5.9328(2) Å



# Single Crystals



1.5466 g and 1.2783 g  
single crystals of  
 $\text{FeTe}_{0.7}\text{Se}_{0.3}$

Deintercalation  
Single Crystal diffractions  
and inelastic neutron  
scattering

# Conclusions

- ◉  $I_2$  reactions successfully removed interstitial Fe
- ◉ Higher amounts of Se lowered amount of interstitial iron
- ◉ Slow cooled is the better method for synthesis of 50:50
- ◉ Will allow us to explore new parts of the phase diagram



# FUTURE WORK



- ◉ Analyze deintercalation through acetonitrile results
- ◉ Neutron Studies  $\text{FeTe}_{0.9}\text{Se}_{0.1}$  after  $\text{I}_2$  deintercalation
- ◉ Complete phase diagram

*From the film "Avatar"*  
Avtr.com

# Thanks to:

- Dr. Efrain E. Rodriguez, Advisor
- Jonny Schear, SHIP student
- Dr. Julie Borchers
- NCNR Staff
- Surf Organizers

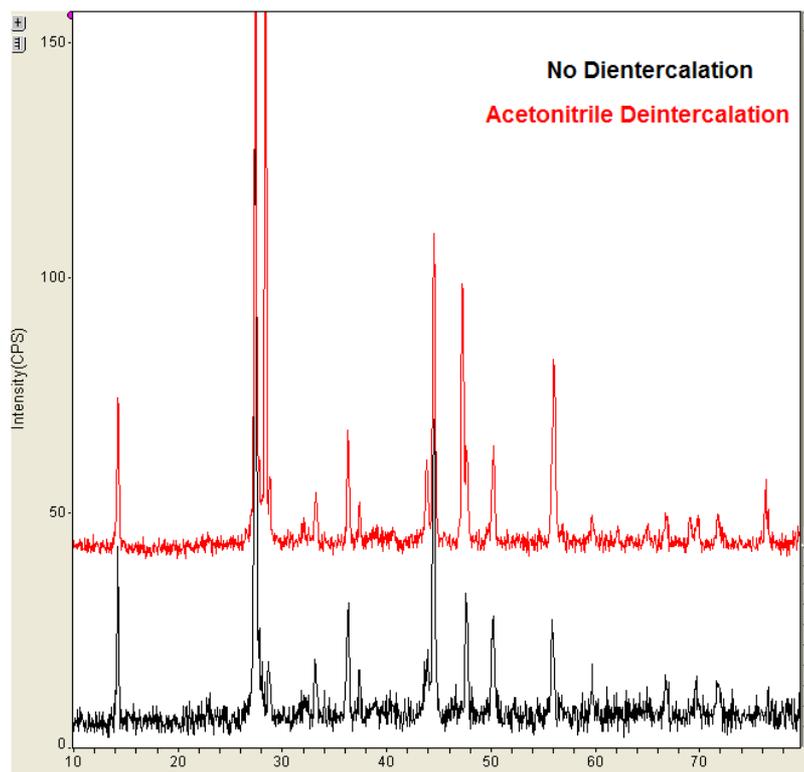




# Acetonitrile Reactions

## Still in progress...

FeTe<sub>0.9</sub>Se<sub>0.1</sub> + Si Standard



FeTe<sub>0.8</sub>Se<sub>0.2</sub> + Si Standard

