# Rotational Ligand Dynamics in Mn[N(CN)<sub>2</sub>]<sub>2</sub>.pyrazine

#### Group C

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 material consisting of transition metal ions linked by polydentate organic ligands

 forms different structures at low and high temperatures

- at 1.3K 3D antiferromagnetic order, monoclinic lattice
- ~ 200K orthorhombic structure
- ~ 408K DSC shows additional phase transition ?



### Why use neutron scattering?

 Have wavelengths comparable to interatomic distances and energies comparable to phonon and intermolecular interaction energies

: obtain information on geometry and time-scale of motions simultaneously

- Interact differently with H and D
- Have a magnetic moment
- Interact directly with the nucleus

### **Disc-Chopper Spectrometer**



## **Measured Scattering**

- Incoherent scattering dominated by protons
- $S_{inc}(Q,\omega)$  is the space and time FT of the selfcorrelation function...probability of finding a particle at time t in position r given that it was at the origin at time t=0
- Fit using model such as diffusional motion of protons among equivalent sites

$$S_{inc}(Q,\omega) = \frac{1}{2} \left( 1 + \frac{\sin[QR]}{QR} \right) \delta(\omega) + \frac{1}{2} \left( 1 - \frac{\sin[QR]}{QR} \right) \frac{1}{\pi} \left( \frac{2\tau}{4 + \tau^2 \omega^2} \right)$$
  
Elastic contribution Inelastic contribution

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#### Results

$$EISF = \frac{I_{elastic}}{I_{total}} = \frac{G_p}{G_p + L_p - (m_p / m_d)L_d}$$





## Conclusion

A thermally activated 2-fold jump motion about the coordinating nitrogen axis takes place which actually starts at 200K. At 408K more pyrazine rings perform this motion and a more open framework structure forms.

Thanks to the organizers!